

**JEE April 2019**

Test Date	08/04/2019
Test Time	9:30 AM - 12:30 PM
Subject	Paper I EH

Section : Physics

**Q.1** A boy's catapult is made of rubber cord which is 42 cm long, with 6 mm diameter of cross-section and of negligible mass. The boy keeps a stone weighing 0.02 kg on it and stretches the cord by 20 cm by applying a constant force. When released, the stone flies off with a velocity of  $20 \text{ ms}^{-1}$ . Neglect the change in the area of cross-section of the cord while stretched. The Young's modulus of rubber is closest to :

- Options
1.  $10^6 \text{ Nm}^{-2}$  ✓
  2.  $10^4 \text{ Nm}^{-2}$
  3.  $10^8 \text{ Nm}^{-2}$
  4.  $10^3 \text{ Nm}^{-2}$

Question Type : **MCQ**Question ID : **41652912702**Option 1 ID : **41652949586**Option 2 ID : **41652949588**Option 3 ID : **41652949587**Option 4 ID : **41652949589**Status : **Answered**Chosen Option : **4**

**Q.2** A thermally insulated vessel contains 150 g of water at  $0^\circ\text{C}$ . Then the air from the vessel is pumped out adiabatically. A fraction of water turns into ice and the rest evaporates at  $0^\circ\text{C}$  itself. The mass of evaporated water will be closest to :

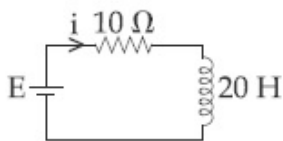
(Latent heat of vaporization of water =  $2.10 \times 10^6 \text{ J kg}^{-1}$  and Latent heat of Fusion of water =  $3.36 \times 10^5 \text{ J kg}^{-1}$ )

- Options
1. 35 g

2. 20 g ✓
3. 130 g
4. 150 g

Question Type : **MCQ**Question ID : **41652912705**Option 1 ID : **41652949600**Option 2 ID : **41652949599**Option 3 ID : **41652949601**Option 4 ID : **41652949598**Status : **Answered**Chosen Option : **2**

- Q.3** A 20 Henry inductor coil is connected to a 10 ohm resistance in series as shown in figure. The time at which rate of dissipation of energy (Joule's heat) across resistance is equal to the rate at which magnetic energy is stored in the inductor, is :



Options

1.  $\frac{1}{2} \ln 2$
2.  $2 \ln 2$  ✓
3.  $\frac{2}{\ln 2}$
4.  $\ln 2$

Question Type : **MCQ**Question ID : **41652912716**Option 1 ID : **41652949642**Option 2 ID : **41652949643**Option 3 ID : **41652949644**Option 4 ID : **41652949645**Status : **Not Answered**Chosen Option : **--**

- Q.4** Radiation coming from transitions  $n=2$  to  $n=1$  of hydrogen atoms fall on  $\text{He}^+$  ions in  $n=1$  and  $n=2$  states. The possible transition of helium ions as they absorb energy from the radiation is :

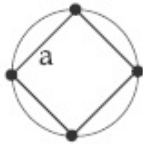
Options

1.  $n=2 \rightarrow n=3$
2.  $n=2 \rightarrow n=4$  ✓
3.  $n=2 \rightarrow n=5$

4.  $n = 1 \rightarrow n = 4$

Question Type : **MCQ**Question ID : **41652912722**Option 1 ID : **41652949668**Option 2 ID : **41652949669**Option 3 ID : **41652949666**Option 4 ID : **41652949667**Status : **Answered**Chosen Option : **2**

- Q.5** Four identical particles of mass  $M$  are located at the corners of a square of side 'a'. What should be their speed if each of them revolves under the influence of others' gravitational field in a circular orbit circumscribing the square ?



Options

1.  $1.35 \sqrt{\frac{GM}{a}}$

2.  $1.21 \sqrt{\frac{GM}{a}}$

3.  $1.41 \sqrt{\frac{GM}{a}}$

4.  $1.16 \sqrt{\frac{GM}{a}}$  ✓

Question Type : **MCQ**Question ID : **41652912701**Option 1 ID : **41652949583**Option 2 ID : **41652949582**Option 3 ID : **41652949584**Option 4 ID : **41652949585**Status : **Answered**Chosen Option : **3**

- Q.6** The wavelength of the carrier waves in a modern optical fiber communication network is close to :

Options 1. 2400 nm

2. 900 nm

3. 600 nm

4. 1500 nm ✓

Question Type : **MCQ**Question ID : **41652912724**Option 1 ID : **41652949676**Option 2 ID : **41652949675**Option 3 ID : **41652949677**Option 4 ID : **41652949674**Status : **Answered**Chosen Option : **3**

**Q.7** A thin circular plate of mass  $M$  and radius  $R$  has its density varying as  $\rho(r) = \rho_0 r$  with  $\rho_0$  as constant and  $r$  is the distance from its center. The moment of Inertia of the circular plate about an axis perpendicular to the plate and passing through its edge is  $I = a MR^2$ . The value of the coefficient  $a$  is :

Options

1.  $\frac{3}{5}$

2.  $\frac{1}{2}$

3.  $\frac{8}{5}$  ✓

4.  $\frac{3}{2}$

Question Type : **MCQ**Question ID : **41652912700**Option 1 ID : **41652949580**Option 2 ID : **41652949581**Option 3 ID : **41652949578**Option 4 ID : **41652949579**Status : **Answered**Chosen Option : **4**

**Q.8** An alternating voltage  $v(t) = 220 \sin 100\pi t$  volt is applied to a purely resistive load of  $50 \Omega$ . The time taken for the current to rise from half of the peak value to the peak value is :

Options 1. 7.2 ms

2. 5 ms

3. 2.2 ms

4. 3.3 ms ✓

Question Type : **MCQ**Question ID : **41652912714**Option 1 ID : **41652949637**Option 2 ID : **41652949636**Option 3 ID : **41652949634**

Option 4 ID : 41652949635

Status : Answered

Chosen Option : 4

**Q.9** The bob of a simple pendulum has mass 2 g and a charge of  $5.0 \mu\text{C}$ . It is at rest in a uniform horizontal electric field of intensity  $2000 \text{ V/m}$ . At equilibrium, the angle that the pendulum makes with the vertical is :  
(take  $g = 10 \text{ m/s}^2$ )

- Options
1.  $\tan^{-1}(0.2)$
  2.  $\tan^{-1}(2.0)$
  3.  $\tan^{-1}(0.5)$  ✓
  4.  $\tan^{-1}(5.0)$

Question Type : MCQ

Question ID : 41652912709

Option 1 ID : 41652949617

Option 2 ID : 41652949616

Option 3 ID : 41652949614

Option 4 ID : 41652949615

Status : Answered

Chosen Option : 3

**Q.10** A circular coil having  $N$  turns and radius  $r$  carries a current  $I$ . It is held in the  $XZ$  plane in a magnetic field  $\hat{B}_i$ . The torque on the coil due to the magnetic field is :

- Options
1.  $B\pi r^2 I N$  ✓
  2.  $\frac{Br^2 I}{\pi N}$
  3.  $\frac{B\pi r^2 I}{N}$
  4. Zero

Question Type : MCQ

Question ID : 41652912713

Option 1 ID : 41652949630

Option 2 ID : 41652949633

Option 3 ID : 41652949631

Option 4 ID : 41652949632

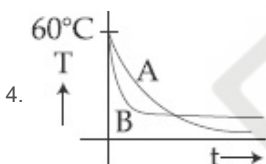
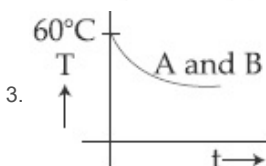
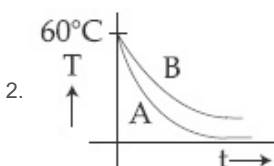
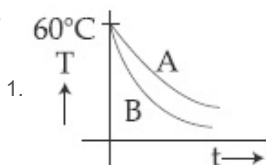
Status : Answered

Chosen Option : 1

Q.11

Two identical beakers A and B contain equal volumes of two different liquids at  $60^\circ\text{C}$  each and left to cool down. Liquid in A has density of  $8 \times 10^2 \text{ kg/m}^3$  and specific heat of  $2000 \text{ J kg}^{-1} \text{ K}^{-1}$  while liquid in B has density of  $10^3 \text{ kg m}^{-3}$  and specific heat of  $4000 \text{ J kg}^{-1} \text{ K}^{-1}$ . Which of the following best describes their temperature versus time graph schematically ? (assume the emissivity of both the beakers to be the same)

Options



Question Type : MCQ

Question ID : 41652912704

Option 1 ID : 41652949594

Option 2 ID : 41652949595

Option 3 ID : 41652949596

Option 4 ID : 41652949597

Status : Not Answered

Chosen Option : --

Q.12 If  $10^{22}$  gas molecules each of mass  $10^{-26} \text{ kg}$  collide with a surface (perpendicular to it) elastically per second over an area  $1 \text{ m}^2$  with a speed  $10^4 \text{ m/s}$ , the pressure exerted by the gas molecules will be of the order of :

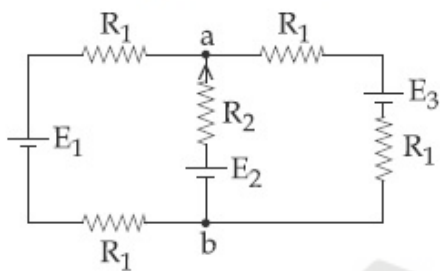
Options 1.  $10^4 \text{ N/m}^2$

2.  $10^{16} \text{ N/m}^2$
3.  $10^8 \text{ N/m}^2$
4.  $10^3 \text{ N/m}^2$  ✓

Question Type : **MCQ**Question ID : **41652912706**Option 1 ID : **41652949603**Option 2 ID : **41652949605**Option 3 ID : **41652949604**Option 4 ID : **41652949602**Status : **Not Answered**

Chosen Option : --

- Q.13** For the circuit shown, with  $R_1 = 1.0 \Omega$ ,  $R_2 = 2.0 \Omega$ ,  $E_1 = 2 \text{ V}$  and  $E_2 = E_3 = 4 \text{ V}$ , the potential difference between the points 'a' and 'b' is approximately ( in V ) :



- Options
1. 3.3 ✓
  2. 2.3
  3. 2.7
  4. 3.7

Question Type : **MCQ**Question ID : **41652912712**Option 1 ID : **41652949627**Option 2 ID : **41652949629**Option 3 ID : **41652949628**Option 4 ID : **41652949626**Status : **Not Answered**

Chosen Option : --

- Q.14** A plane electromagnetic wave travels in free space along the  $x$ -direction. The electric field component of the wave at a particular point of space and time is  $E = 6 \text{ Vm}^{-1}$  along  $y$ -direction. Its corresponding magnetic field component,  $B$  would be :

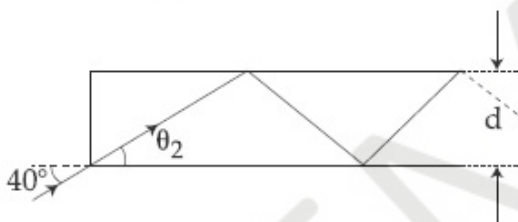
- Options
1.  $2 \times 10^{-8} \text{ T}$  along  $z$ -direction ✓
  2.  $2 \times 10^{-8} \text{ T}$  along  $y$ -direction

3.  $6 \times 10^{-8}$  T along z-direction
4.  $6 \times 10^{-8}$  T along x-direction

Question Type : **MCQ**Question ID : **41652912717**Option 1 ID : **41652949646**Option 2 ID : **41652949648**Option 3 ID : **41652949649**Option 4 ID : **41652949647**Status : **Not Answered**

Chosen Option : --

- Q.15** In figure, the optical fiber is  $l=2$  m long and has a diameter of  $d=20 \mu\text{m}$ . If a ray of light is incident on one end of the fiber at angle  $\theta_1=40^\circ$ , the number of reflections it makes before emerging from the other end is close to :  
(refractive index of fiber is 1.31 and  $\sin 40^\circ = 0.64$ )



- Options
1. 55000
  2. 57000 ✓
  3. 45000
  4. 66000

Question Type : **MCQ**Question ID : **41652912720**Option 1 ID : **41652949660**Option 2 ID : **41652949658**Option 3 ID : **41652949661**Option 4 ID : **41652949659**Status : **Not Answered**

Chosen Option : --

- Q.16** Two particles move at right angle to each other. Their de Broglie wavelengths are  $\lambda_1$  and  $\lambda_2$  respectively. The particles suffer perfectly *inelastic* collision. The de Broglie wavelength  $\lambda$ , of the final particle, is given by :

Options



$$1. \frac{2}{\lambda} = \frac{1}{\lambda_1} + \frac{1}{\lambda_2}$$

$$2. \lambda = \frac{\lambda_1 + \lambda_2}{2}$$

$$3. \lambda = \sqrt{\lambda_1 \lambda_2}$$

$$4. \frac{1}{\lambda^2} = \frac{1}{\lambda_1^2} + \frac{1}{\lambda_2^2} \quad \checkmark$$

Question Type : **MCQ**

Question ID : **41652912721**

Option 1 ID : **41652949664**

Option 2 ID : **41652949662**

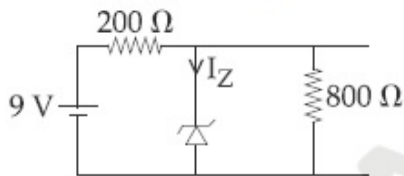
Option 3 ID : **41652949663**

Option 4 ID : **41652949665**

Status : **Not Answered**

Chosen Option : --

**Q.17** The reverse breakdown voltage of a Zener diode is 5.6 V in the given circuit.



The current  $I_z$  through the Zener is :

- Options
1. 10 mA
  2. 7 mA
  3. 17 mA
  4. 15 mA

Question Type : **MCQ**

Question ID : **41652912723**

Option 1 ID : **41652949671**

Option 2 ID : **41652949672**

Option 3 ID : **41652949673**

Option 4 ID : **41652949670**

Status : **Not Answered**

Chosen Option : --

**Q.18** An upright object is placed at a distance of 40 cm in front of a convergent lens of focal length 20 cm. A convergent mirror of focal length 10 cm is placed at a distance of 60 cm on the other side of the lens. The position and size of the final image will be :

- Options
1. 40 cm from the convergent lens, twice the size of the object
  2. 20 cm from the convergent mirror, twice the size of the object
  3. 40 cm from the convergent mirror, same size as the object
  4. 20 cm from the convergent mirror, same size as the object ✓

Question Type : **MCQ**

Question ID : **41652912718**

Option 1 ID : **41652949653**

Option 2 ID : **41652949652**

Option 3 ID : **41652949650**

Option 4 ID : **41652949651**

Status : **Answered**

Chosen Option : **4**

Q.19 In an interference experiment the ratio of amplitudes of coherent waves is  $\frac{a_1}{a_2} = \frac{1}{3}$ .

The ratio of maximum and minimum intensities of fringes will be :

- Options
1. 4 ✓
  2. 9
  3. 2
  4. 18

Question Type : **MCQ**

Question ID : **41652912719**

Option 1 ID : **41652949655**

Option 2 ID : **41652949656**

Option 3 ID : **41652949654**

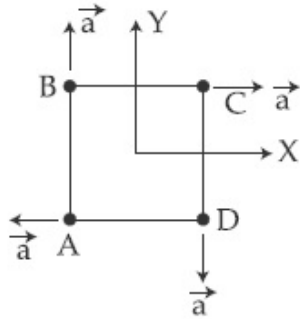
Option 4 ID : **41652949657**

Status : **Not Answered**

Chosen Option : **--**

Q.20

Four particles A, B, C and D with masses  $m_A = m$ ,  $m_B = 2m$ ,  $m_C = 3m$  and  $m_D = 4m$  are at the corners of a square. They have accelerations of equal magnitude with directions as shown. The acceleration of the centre of mass of the particles is :



- Options
1.  $\frac{a}{5}(\hat{i} + \hat{j})$
  2.  $\frac{a}{5}(\hat{i} - \hat{j})$  ✓
  3.  $a(\hat{i} + \hat{j})$
  4. Zero

Question Type : MCQ

Question ID : 41652912698

Option 1 ID : 41652949572

Option 2 ID : 41652949573

Option 3 ID : 41652949571

Option 4 ID : 41652949570

Status : Answered

Chosen Option : 2

Q.21 A  $200\ \Omega$  resistor has a certain color code. If one replaces the red color by green in the code, the new resistance will be :

- Options
1.  $300\ \Omega$
  2.  $100\ \Omega$
  3.  $400\ \Omega$
  4.  $500\ \Omega$  ✓

Question Type : MCQ

Question ID : 41652912725

Option 1 ID : 41652949679

Option 2 ID : 41652949678

Option 3 ID : 41652949680

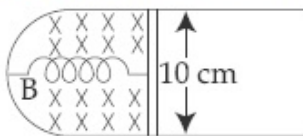
Option 4 ID : 41652949681

Status : Answered

Chosen Option : 4

Q.22

A thin strip 10 cm long is on a U shaped wire of negligible resistance and it is connected to a spring of spring constant  $0.5 \text{ Nm}^{-1}$  (see figure). The assembly is kept in a uniform magnetic field of 0.1 T. If the strip is pulled from its equilibrium position and released, the number of oscillations it performs before its amplitude decreases by a factor of  $e$  is  $N$ . If the mass of the strip is 50 grams, its resistance  $10 \Omega$  and air drag negligible,  $N$  will be close to :



Options 1. 1000

2. 5000 ✓

3. 10000

4. 50000

Question Type : MCQ

Question ID : 41652912715

Option 1 ID : 41652949638

Option 2 ID : 41652949639

Option 3 ID : 41652949640

Option 4 ID : 41652949641

Status : Not Answered

Chosen Option : --

Q.23

A steel wire having a radius of 2.0 mm, carrying a load of 4 kg, is hanging from a ceiling. Given that  $g = 3.1 \pi \text{ ms}^{-2}$ , what will be the tensile stress that would be developed in the wire ?

Options 1.  $5.2 \times 10^6 \text{ Nm}^{-2}$ 2.  $6.2 \times 10^6 \text{ Nm}^{-2}$ 3.  $4.8 \times 10^6 \text{ Nm}^{-2}$ 4.  $3.1 \times 10^6 \text{ Nm}^{-2}$  ✓

Question Type : MCQ

Question ID : 41652912707

Option 1 ID : 41652949606

Option 2 ID : 41652949608

Option 3 ID : 41652949609

Option 4 ID : 41652949607

Status : Answered

Chosen Option : 4

Q.24 Ship A is sailing towards north-east with velocity  $\vec{v} = 30\hat{i} + 50\hat{j}$  km/hr where  $\hat{i}$  points east and  $\hat{j}$ , north. Ship B is at a distance of 80 km east and 150 km north of Ship A and is sailing towards west at 10 km/hr. A will be at minimum distance from B in :

- Options
1. 4.2 hrs.
  2. 3.2 hrs.
  3. 2.6 hrs. ✓
  4. 2.2 hrs.

Question Type : MCQ

Question ID : 41652912697

Option 1 ID : 41652949566

Option 2 ID : 41652949569

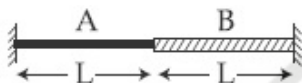
Option 3 ID : 41652949568

Option 4 ID : 41652949567

Status : Not Answered

Chosen Option : --

Q.25



A wire of length  $2L$ , is made by joining two wires A and B of same length but different radii  $r$  and  $2r$  and made of the same material. It is vibrating at a frequency such that the joint of the two wires forms a node. If the number of antinodes in wire A is  $p$  and that in B is  $q$  then the ratio  $p : q$  is :

- Options
1. 3 : 5
  2. 4 : 9
  3. 1 : 4
  4. 1 : 2 ✓

Question Type : MCQ

Question ID : 41652912708

Option 1 ID : 41652949611

Option 2 ID : 41652949612

Option 3 ID : 41652949613

Option 4 ID : 41652949610

Status : Answered

Chosen Option : 4

Q.26 A solid conducting sphere, having a charge  $Q$ , is surrounded by an uncharged conducting hollow spherical shell. Let the potential difference between the surface of the solid sphere and that of the outer surface of the hollow shell be  $V$ . If the shell is now given a charge of  $-4Q$ , the new potential difference between the same two surfaces is :

- Options
1.  $-2V$
  2.  $2V$
  3.  $V$  ✓
  4.  $4V$

Question Type : MCQ

Question ID : 41652912710

Option 1 ID : 41652949620

Option 2 ID : 41652949619

Option 3 ID : 41652949618

Option 4 ID : 41652949621

Status : Not Answered

Chosen Option : --

Q.27 Water from a pipe is coming at a rate of 100 liters per minute. If the radius of the pipe is 5 cm, the Reynolds number for the flow is of the order of : (density of water =  $1000 \text{ kg/m}^3$ , coefficient of viscosity of water =  $1 \text{ mPa s}$ )

- Options
1.  $10^2$
  2.  $10^4$  ✓
  3.  $10^3$
  4.  $10^6$

Question Type : MCQ

Question ID : 41652912703

Option 1 ID : 41652949593

Option 2 ID : 41652949590

Option 3 ID : 41652949591

Option 4 ID : 41652949592

Status : Answered

Chosen Option : 2

Q.28 In SI units, the dimensions of  $\sqrt{\frac{\epsilon_0}{\mu_0}}$  is :

- Options
1.  $\text{AT}^2\text{M}^{-1}\text{L}^{-1}$

2.  $A^2T^3M^{-1}L^{-2}$  ✓
3.  $A^{-1}TML^3$
4.  $AT^{-3}ML^{3/2}$

Question Type : MCQ

Question ID : 41652912696

Option 1 ID : 41652949562

Option 2 ID : 41652949563

Option 3 ID : 41652949564

Option 4 ID : 41652949565

Status : Answered

Chosen Option : 2

**Q.29** Voltage rating of a parallel plate capacitor is 500 V. Its dielectric can withstand a maximum electric field of  $10^6$  V/m. The plate area is  $10^{-4}$  m<sup>2</sup>. What is the dielectric constant if the capacitance is 15 pF ?  
(given  $\epsilon_0 = 8.86 \times 10^{-12}$  C<sup>2</sup>/Nm<sup>2</sup>)

Options 1. 3.8

2. 8.5 ✓
3. 4.5
4. 6.2

Question Type : MCQ

Question ID : 41652912711

Option 1 ID : 41652949623

Option 2 ID : 41652949625

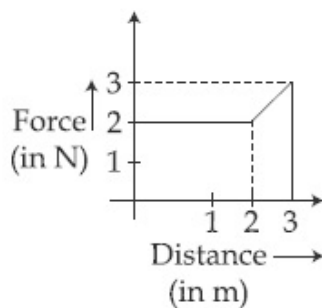
Option 3 ID : 41652949624

Option 4 ID : 41652949622

Status : Not Answered

Chosen Option : --

**Q.30** A particle moves in one dimension from rest under the influence of a force that varies with the distance travelled by the particle as shown in the figure. The kinetic energy of the particle after it has travelled 3 m is :



Options 1. 4 J

2. 2.5 J

3. 6.5 J ✓

4. 5 J

Question Type : MCQ

Question ID : 41652912699

Option 1 ID : 41652949576

Option 2 ID : 41652949577

Option 3 ID : 41652949575

Option 4 ID : 41652949574

Status : Not Answered

Chosen Option : --

Section : Chemistry

Q.1 The quantum number of four electrons are given below :

I.  $n=4, l=2, m_l=-2, m_s=-\frac{1}{2}$ II.  $n=3, l=2, m_l=1, m_s=+\frac{1}{2}$ III.  $n=4, l=1, m_l=0, m_s=+\frac{1}{2}$ IV.  $n=3, l=1, m_l=1, m_s=-\frac{1}{2}$ 

The correct order of their increasing energies will be :

Options 1. I &lt; III &lt; II &lt; IV

2. IV &lt; II &lt; III &lt; I ✓

3. I &lt; II &lt; III &lt; IV

4. IV &lt; III &lt; II &lt; I

Question Type : MCQ

Question ID : 41652912748

Option 1 ID : 41652949771

Option 2 ID : 41652949773

Option 3 ID : 41652949770

Option 4 ID : 41652949772

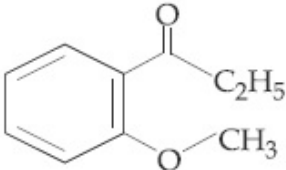
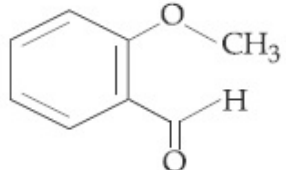
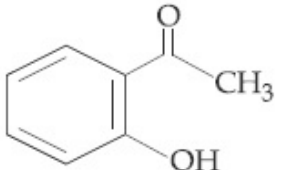
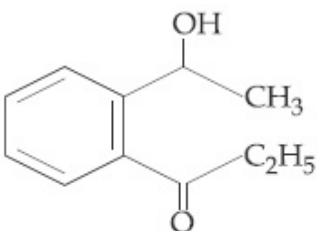
Status : Answered

Chosen Option : 2

Q.2 An organic compound neither reacts with neutral ferric chloride solution nor with Fehling solution. It however, reacts with Grignard reagent and gives positive iodoform test. The compound is :

Options



1. 
2. 
3. 
4. 

✓

Question Type : **MCQ**Question ID : **41652912726**Option 1 ID : **41652949684**Option 2 ID : **41652949683**Option 3 ID : **41652949682**Option 4 ID : **41652949685**Status : **Not Answered**

Chosen Option : --

**Q.3** Maltose on treatment with dilute HCl gives :

- Options
1. D-Glucose ✓
  2. D-Fructose
  3. D-Galactose
  4. D-Glucose and D-Fructose

Question Type : **MCQ**Question ID : **41652912727**Option 1 ID : **41652949689**Option 2 ID : **41652949688**Option 3 ID : **41652949686**Option 4 ID : **41652949687**Status : **Not Answered**

Chosen Option : --

**Q.4**

For silver,  $C_p(\text{J K}^{-1} \text{mol}^{-1}) = 23 + 0.01T$ . If the temperature (T) of 3 moles of silver is raised from 300 K to 1000 K at 1 atm pressure, the value of  $\Delta H$  will be close to :

- Options
1. 16 kJ
  2. 62 kJ ✓
  3. 13 kJ
  4. 21 kJ

Question Type : MCQ

Question ID : 41652912749

Option 1 ID : 41652949776

Option 2 ID : 41652949777

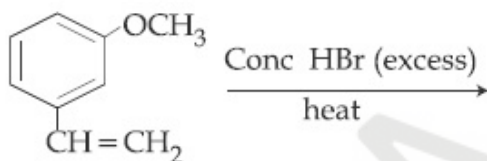
Option 3 ID : 41652949774

Option 4 ID : 41652949775

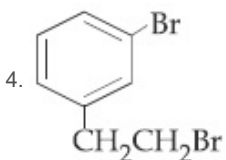
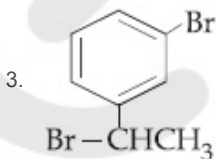
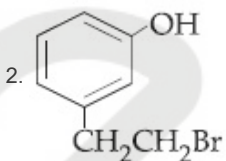
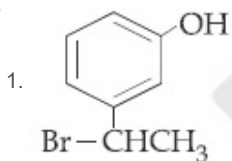
Status : Answered

Chosen Option : 2

Q.5 The major product of the following reaction is :



Options



Question Type : MCQ

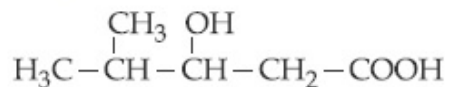
Question ID : 41652912730

Option 1 ID : 41652949699

Option 2 ID : 41652949698

Option 3 ID : 41652949701  
 Option 4 ID : 41652949700  
 Status : Answered  
 Chosen Option : 1

Q.6 The IUPAC name of the following compound is :



- Options
1. 4-Methyl-3-hydroxypentanoic acid
  2. 4,4-Dimethyl-3-hydroxybutanoic acid
  3. 2-Methyl-3-hydroxypentan-5-oic acid
  4. 3-Hydroxy-4-methylpentanoic acid ✓

Question Type : MCQ  
 Question ID : 41652912733  
 Option 1 ID : 41652949711  
 Option 2 ID : 41652949712  
 Option 3 ID : 41652949713  
 Option 4 ID : 41652949710  
 Status : Answered  
 Chosen Option : 4

Q.7 Which of the following amines can be prepared by Gabriel phthalimide reaction ?

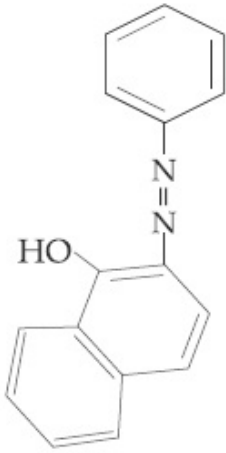
- Options
1. t-butylamine
  2. n-butylamine ✓
  3. triethylamine
  4. neo-pentylamine

Question Type : MCQ  
 Question ID : 41652912732  
 Option 1 ID : 41652949708  
 Option 2 ID : 41652949707  
 Option 3 ID : 41652949709  
 Option 4 ID : 41652949706  
 Status : Answered  
 Chosen Option : 3

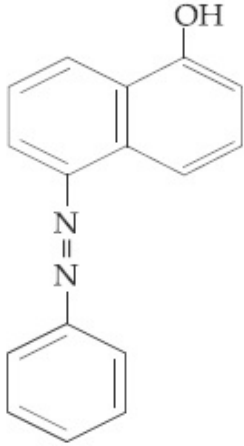
Q.8 Coupling of benzene diazonium chloride with 1-naphthol in alkaline medium will give :

Options

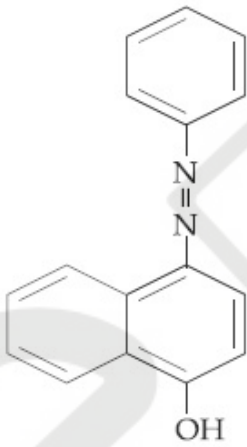
1.



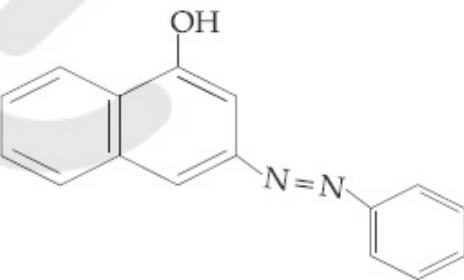
2.



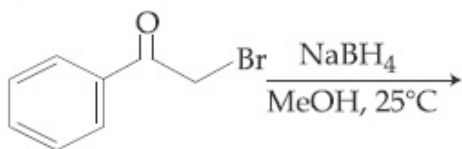
3.



4.

Question Type : **MCQ**Question ID : **41652912728**Option 1 ID : **41652949691**Option 2 ID : **41652949693**Option 3 ID : **41652949690**Option 4 ID : **41652949692**

Q.9 The major product of the following reaction is :



Options

1. ✓
- 2.
- 3.
- 4.

Question Type : MCQ

Question ID : 41652912731

Option 1 ID : 41652949702

Option 2 ID : 41652949704

Option 3 ID : 41652949705

Option 4 ID : 41652949703

Status : Not Answered

Chosen Option : --

Q.10 An organic compound 'X' showing the following solubility profile is :

water	→ insoluble
'X'	
5% HCl	→ insoluble
10% NaOH	→ soluble
10% NaHCO <sub>3</sub>	→ insoluble

- Options
1. m-Cresol ✓
  2. o-Toluidine
  3. Oleic acid

## 4. Benzamide

Question Type : MCQ

Question ID : 41652912734

Option 1 ID : 41652949715

Option 2 ID : 41652949716

Option 3 ID : 41652949714

Option 4 ID : 41652949717

Status : Not Answered

Chosen Option : --

Q.11 100 mL of a water sample contains 0.81 g of calcium bicarbonate and 0.73 g of magnesium bicarbonate. The hardness of this water sample expressed in terms of equivalents of  $\text{CaCO}_3$  is :

(molar mass of calcium bicarbonate is  $162 \text{ g mol}^{-1}$  and magnesium bicarbonate is  $146 \text{ g mol}^{-1}$ )

- Options
1. 100 ppm
  2. 1,000 ppm
  3. 5,000 ppm
  4. 10,000 ppm ✓

Question Type : MCQ

Question ID : 41652912738

Option 1 ID : 41652949732

Option 2 ID : 41652949730

Option 3 ID : 41652949733

Option 4 ID : 41652949731

Status : Not Answered

Chosen Option : --

Q.12 Diborane ( $\text{B}_2\text{H}_6$ ) reacts independently with  $\text{O}_2$  and  $\text{H}_2\text{O}$  to produce, respectively :

- Options
1.  $\text{HBO}_2$  and  $\text{H}_3\text{BO}_3$
  2.  $\text{H}_3\text{BO}_3$  and  $\text{B}_2\text{O}_3$
  3.  $\text{B}_2\text{O}_3$  and  $[\text{BH}_4]^-$
  4.  $\text{B}_2\text{O}_3$  and  $\text{H}_3\text{BO}_3$  ✓

Question Type : MCQ

Question ID : 41652912740

Option 1 ID : 41652949739

Option 2 ID : 41652949738

Option 3 ID : 41652949741

Option 4 ID : 41652949740

Status : **Answered**  
Chosen Option : 4

**Q.13** The size of the iso-electronic species  $\text{Cl}^-$ ,  $\text{Ar}$  and  $\text{Ca}^{2+}$  is affected by :

- Options
1. nuclear charge ✓
  2. azimuthal quantum number of valence shell
  3. electron-electron interaction in the outer orbitals
  4. Principal quantum number of valence shell

Question Type : **MCQ**  
Question ID : **41652912736**  
Option 1 ID : **41652949723**  
Option 2 ID : **41652949724**  
Option 3 ID : **41652949725**  
Option 4 ID : **41652949722**  
Status : **Answered**  
Chosen Option : 1

**Q.14** Element 'B' forms ccp structure and 'A' occupies half of the octahedral voids, while oxygen atoms occupy all the tetrahedral voids. The structure of bimetallic oxide is :

- Options
1.  $\text{A}_2\text{BO}_4$
  2.  $\text{AB}_2\text{O}_4$  ✓
  3.  $\text{A}_2\text{B}_2\text{O}$
  4.  $\text{A}_4\text{B}_2\text{O}$

Question Type : **MCQ**  
Question ID : **41652912747**  
Option 1 ID : **41652949766**  
Option 2 ID : **41652949767**  
Option 3 ID : **41652949769**  
Option 4 ID : **41652949768**  
Status : **Answered**  
Chosen Option : 2

**Q.15** With respect to an ore, Ellingham diagram helps to predict the feasibility of its

- Options
1. Electrolysis
  2. Zone refining
  3. Vapour phase refining

## 4. Thermal reduction ✓

Question Type : MCQ

Question ID : 41652912737

Option 1 ID : 41652949727

Option 2 ID : 41652949728

Option 3 ID : 41652949729

Option 4 ID : 41652949726

Status : Answered

Chosen Option : 4

Q.16

For the reaction  $2A + B \rightarrow C$ , the values of initial rate at different reactant concentrations are given in the table below. The rate law for the reaction is :

[A] (mol L <sup>-1</sup> )	[B] (mol L <sup>-1</sup> )	Initial Rate (mol L <sup>-1</sup> s <sup>-1</sup> )
0.05	0.05	0.045
0.10	0.05	0.090
0.20	0.10	0.72

Options 1. Rate = k[A]<sup>2</sup>[B]2. Rate = k[A]<sup>2</sup>[B]<sup>2</sup>

3. Rate = k[A][B]

4. Rate = k[A][B]<sup>2</sup> ✓

Question Type : MCQ

Question ID : 41652912754

Option 1 ID : 41652949795

Option 2 ID : 41652949797

Option 3 ID : 41652949794

Option 4 ID : 41652949796

Status : Answered

Chosen Option : 4

Q.17 The correct order of the spin-only magnetic moment of metal ions in the following low-spin complexes,  $[V(CN)_6]^{4-}$ ,  $[Fe(CN)_6]^{4-}$ ,  $[Ru(NH_3)_6]^{3+}$ , and  $[Cr(NH_3)_6]^{2+}$ , is :

Options 1.  $V^{2+} > Cr^{2+} > Ru^{3+} > Fe^{2+}$  ✓2.  $Cr^{2+} > V^{2+} > Ru^{3+} > Fe^{2+}$ 3.  $Cr^{2+} > Ru^{3+} > Fe^{2+} > V^{2+}$ 4.  $V^{2+} > Ru^{3+} > Cr^{2+} > Fe^{2+}$

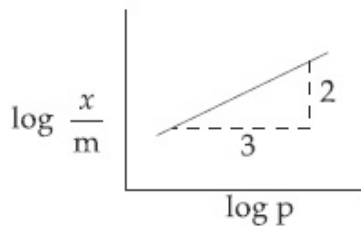


Question Type : **MCQ**Question ID : **41652912743**Option 1 ID : **41652949750**Option 2 ID : **41652949753**Option 3 ID : **41652949752**Option 4 ID : **41652949751**Status : **Answered**Chosen Option : **2**

Q.18 Adsorption of a gas follows Freundlich adsorption isotherm.  $x$  is the mass of the gas adsorbed on mass  $m$  of the adsorbent.

The plot of  $\log \frac{x}{m}$  versus  $\log p$  is shown in

the given graph.  $\frac{x}{m}$  is proportional to :



- Options
1.  $p^3$
  2.  $p^2$
  3.  $p^{2/3}$  ✓
  4.  $p^{3/2}$

Question Type : **MCQ**Question ID : **41652912755**Option 1 ID : **41652949801**Option 2 ID : **41652949798**Option 3 ID : **41652949800**Option 4 ID : **41652949799**Status : **Answered**Chosen Option : **3**

Q.19 Which is wrong with respect to our responsibility as a human being to protect our environment ?

- Options
1. Using plastic bags. ✓
  2. Restricting the use of vehicles
  3. Avoiding the use of floodlighted facilities.
  4. Setting up compost tin in gardens.

Question Type : **MCQ**Question ID : **41652912744**Option 1 ID : **41652949755**Option 2 ID : **41652949756**Option 3 ID : **41652949757**Option 4 ID : **41652949754**Status : **Answered**Chosen Option : **1**

**Q.20** In order to oxidise a mixture of one mole of each of  $\text{FeC}_2\text{O}_4$ ,  $\text{Fe}_2(\text{C}_2\text{O}_4)_3$ ,  $\text{FeSO}_4$  and  $\text{Fe}_2(\text{SO}_4)_3$  in acidic medium, the number of moles of  $\text{KMnO}_4$  required is :

Options 1. 1

2. 2 ✓

3. 3

4. 1.5

Question Type : **MCQ**Question ID : **41652912746**Option 1 ID : **41652949762**Option 2 ID : **41652949764**Option 3 ID : **41652949765**Option 4 ID : **41652949763**Status : **Answered**Chosen Option : **2**

**Q.21** Given that  $E_{\text{O}_2/\text{H}_2\text{O}}^\ominus = +1.23 \text{ V}$ ;

$$E_{\text{S}_2\text{O}_8^{2-}/\text{SO}_4^{2-}}^\ominus = 2.05 \text{ V}$$

$$E_{\text{Br}_2/\text{Br}^-}^\ominus = +1.09 \text{ V};$$

$$E_{\text{Au}^{3+}/\text{Au}}^\ominus = +1.4 \text{ V}$$

The strongest oxidizing agent is :

Options 1.  $\text{O}_2$ 2.  $\text{S}_2\text{O}_8^{2-}$  ✓3.  $\text{Br}_2$ 4.  $\text{Au}^{3+}$ Question Type : **MCQ**Question ID : **41652912753**Option 1 ID : **41652949793**Option 2 ID : **41652949790**Option 3 ID : **41652949792**Option 4 ID : **41652949791**Status : **Answered**

**Q.22** The lanthanide ion that would show colour is :

- Options
1.  $\text{Lu}^{3+}$
  2.  $\text{La}^{3+}$
  3.  $\text{Gd}^{3+}$
  4.  $\text{Sm}^{3+}$  ✓

Question Type : **MCQ**

Question ID : **41652912741**

Option 1 ID : **41652949745**

Option 2 ID : **41652949744**

Option 3 ID : **41652949742**

Option 4 ID : **41652949743**

Status : **Answered**

Chosen Option : **2**

**Q.23** In the following compounds, the decreasing order of basic strength will be :

- Options
1.  $\text{C}_2\text{H}_5\text{NH}_2 > \text{NH}_3 > (\text{C}_2\text{H}_5)_2\text{NH}$
  2.  $(\text{C}_2\text{H}_5)_2\text{NH} > \text{NH}_3 > \text{C}_2\text{H}_5\text{NH}_2$
  3.  $\text{NH}_3 > \text{C}_2\text{H}_5\text{NH}_2 > (\text{C}_2\text{H}_5)_2\text{NH}$
  4.  $(\text{C}_2\text{H}_5)_2\text{NH} > \text{C}_2\text{H}_5\text{NH}_2 > \text{NH}_3$  ✓

Question Type : **MCQ**

Question ID : **41652912735**

Option 1 ID : **41652949721**

Option 2 ID : **41652949720**

Option 3 ID : **41652949719**

Option 4 ID : **41652949718**

Status : **Answered**

Chosen Option : **4**

**Q.24** Assertion : Ozone is destroyed by CFCs in the upper stratosphere.

Reason : Ozone holes increase the amount of UV radiation reaching the earth.

- Options
1. Assertion and reason are correct, but the reason is not the explanation for the assertion. ✓
  2. Assertion is false, but the reason is correct.
  3. Assertion and reason are incorrect.

Assertion and reason are both correct,

4. and the reason is the correct explanation for the assertion.

Question Type : **MCQ**

Question ID : **41652912745**

Option 1 ID : **41652949758**

Option 2 ID : **41652949759**

Option 3 ID : **41652949761**

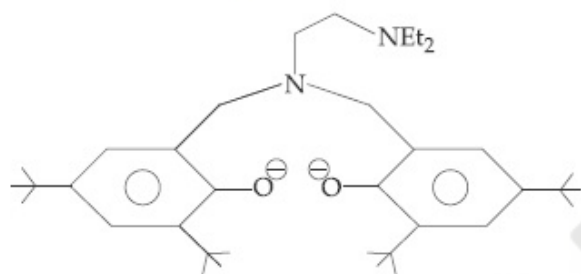
Option 4 ID : **41652949760**

Status : **Answered**

Chosen Option : **4**

Q.25

The following ligand is :



- Options
1. tridentate
  2. hexadentate
  3. bidentate
  4. tetradentate



Question Type : **MCQ**

Question ID : **41652912742**

Option 1 ID : **41652949747**

Option 2 ID : **41652949749**

Option 3 ID : **41652949746**

Option 4 ID : **41652949748**

Status : **Answered**

Chosen Option : **4**

Q.26

The correct order of hydration enthalpies of alkali metal ions is :

- Options
1.  $\text{Na}^+ > \text{Li}^+ > \text{K}^+ > \text{Rb}^+ > \text{Cs}^+$
  2.  $\text{Na}^+ > \text{Li}^+ > \text{K}^+ > \text{Cs}^+ > \text{Rb}^+$
  3.  $\text{Li}^+ > \text{Na}^+ > \text{K}^+ > \text{Rb}^+ > \text{Cs}^+$  ✓
  4.  $\text{Li}^+ > \text{Na}^+ > \text{K}^+ > \text{Cs}^+ > \text{Rb}^+$

Question Type : **MCQ**

Question ID : **41652912739**

Option 1 ID : **41652949736**

Option 2 ID : **41652949737**

Option 3 ID : **41652949734**

Option 4 ID : **41652949735**

Q.27 Which one of the following equations does not correctly represent the first law of thermodynamics for the given processes involving an ideal gas ? (Assume non-expansion work is zero)

- Options
1. Isochoric process :  $\Delta U = q$
  2. Isothermal process :  $q = -w$
  3. Cyclic process :  $q = -w$
  4. Adiabatic process :  $\Delta U = -w$  ✓

Question Type : MCQ

Question ID : 41652912750

Option 1 ID : 41652949780

Option 2 ID : 41652949778

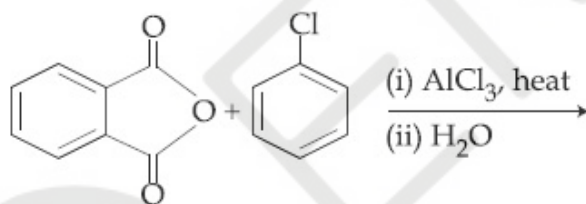
Option 3 ID : 41652949779

Option 4 ID : 41652949781

Status : Answered

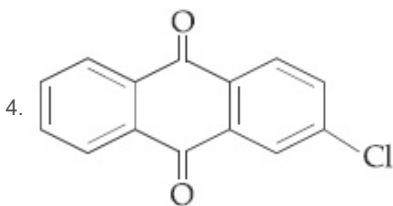
Chosen Option : 4

Q.28 The major product of the following reaction is :



Options

- 1.
2. ✓
- 3.



Question Type : **MCQ**

Question ID : **41652912729**

Option 1 ID : **41652949697**

Option 2 ID : **41652949695**

Option 3 ID : **41652949696**

Option 4 ID : **41652949694**

Status : **Answered**

Chosen Option : **1**

**Q.29** The vapour pressures of pure liquids A and B are 400 and 600 mmHg, respectively at 298 K. On mixing the two liquids, the sum of their initial volumes is equal to the volume of the final mixture. The mole fraction of liquid B is 0.5 in the mixture. The vapour pressure of the final solution, the mole fractions of components A and B in vapour phase, respectively are :

- Options
1. 500 mmHg, 0.5, 0.5
  2. 450 mmHg, 0.4, 0.6
  3. 450 mmHg, 0.5, 0.5
  4. 500 mmHg, 0.4, 0.6 ✓

Question Type : **MCQ**

Question ID : **41652912751**

Option 1 ID : **41652949784**

Option 2 ID : **41652949785**

Option 3 ID : **41652949783**

Option 4 ID : **41652949782**

Status : **Not Answered**

Chosen Option : **--**

**Q.30** If solubility product of  $Zr_3(PO_4)_4$  is denoted by  $K_{sp}$  and its molar solubility is denoted by  $S$ , then which of the following relation between  $S$  and  $K_{sp}$  is correct ?

Options

1.  $S = \left(\frac{K_{sp}}{144}\right)^{1/6}$

$$2. S = \left( \frac{K_{sp}}{6912} \right)^{1/7} \quad \checkmark$$

$$3. S = \left( \frac{K_{sp}}{929} \right)^{1/9}$$

$$4. S = \left( \frac{K_{sp}}{216} \right)^{1/7}$$

Question Type : **MCQ**Question ID : **41652912752**Option 1 ID : **41652949786**Option 2 ID : **41652949788**Option 3 ID : **41652949787**Option 4 ID : **41652949789**Status : **Not Answered**

Chosen Option : --

Section : **Mathematics**

**Q.1** The length of the perpendicular from the point  $(2, -1, 4)$  on the straight line,

$$\frac{x+3}{10} = \frac{y-2}{-7} = \frac{z}{1} \text{ is :}$$

- Options
1. greater than 3 but less than 4
  2. greater than 4
  3. less than 2
  4. greater than 2 but less than 3

Question Type : **MCQ**Question ID : **41652912778**Option 1 ID : **41652949892**Option 2 ID : **41652949893**Option 3 ID : **41652949890**Option 4 ID : **41652949891**Status : **Answered**

Chosen Option : 1

**Q.2** The shortest distance between the line  $y = x$  and the curve  $y^2 = x - 2$  is :

- Options
1.  $\frac{7}{4\sqrt{2}}$
  2.  $\frac{7}{8}$
  3.  $\frac{11}{4\sqrt{2}}$
  4. 2

Question Type : MCQ

Question ID : 41652912776

Option 1 ID : 41652949883

Option 2 ID : 41652949884

Option 3 ID : 41652949882

Option 4 ID : 41652949885

Status : Not Answered

Chosen Option : --

Q.3 Let  $y = y(x)$  be the solution of the differential equation,  $(x^2 + 1)^2 \frac{dy}{dx} + 2x(x^2 + 1)y = 1$  such

that  $y(0) = 0$ . If  $\sqrt{a} y(1) = \frac{\pi}{32}$ , then the value of 'a' is :

Options

1.  $\frac{1}{16}$  ✓
2.  $\frac{1}{2}$
3.  $\frac{1}{4}$
4. 1

Question Type : MCQ

Question ID : 41652912772

Option 1 ID : 41652949866

Option 2 ID : 41652949867

Option 3 ID : 41652949869

Option 4 ID : 41652949868

Status : Answered

Chosen Option : 4

Q.4 All possible numbers are formed using the digits 1, 1, 2, 2, 2, 2, 3, 4, 4 taken all at a time. The number of such numbers in which the odd digits occupy even places is :

Options

1. 175
2. 162
3. 180 ✓
4. 160

Question Type : MCQ

Question ID : 41652912761

Option 1 ID : 41652949824

Option 2 ID : 41652949823

Option 3 ID : 41652949825

Option 4 ID : 41652949822

Status : Answered



Q.5

If  $\alpha = \cos^{-1}\left(\frac{3}{5}\right)$ ,  $\beta = \tan^{-1}\left(\frac{1}{3}\right)$ , where

$0 < \alpha, \beta < \frac{\pi}{2}$ , then  $\alpha - \beta$  is equal to :

Options

1.  $\tan^{-1}\left(\frac{9}{14}\right)$

2.  $\cos^{-1}\left(\frac{9}{5\sqrt{10}}\right)$

3.  $\sin^{-1}\left(\frac{9}{5\sqrt{10}}\right)$  ✓

4.  $\tan^{-1}\left(\frac{9}{5\sqrt{10}}\right)$

Question Type : MCQ

Question ID : 41652912784

Option 1 ID : 41652949914

Option 2 ID : 41652949916

Option 3 ID : 41652949915

Option 4 ID : 41652949917

Status : Answered

Chosen Option : 3

Q.6

If  $f(x) = \log_e\left(\frac{1-x}{1+x}\right)$ ,  $|x| < 1$ , then

$f\left(\frac{2x}{1+x^2}\right)$  is equal to :

Options 1.  $(f(x))^2$

2.  $2f(x^2)$

3.  $-2f(x)$

4.  $2f(x)$  ✓

Question Type : MCQ

Question ID : 41652912756

Option 1 ID : 41652949805

Option 2 ID : 41652949802

Option 3 ID : 41652949803

Option 4 ID : 41652949804

Status : Answered

Chosen Option : 4

Q.7

The equation of a plane containing the line of intersection of the planes  $2x - y - 4 = 0$  and  $y + 2z - 4 = 0$  and passing through the point  $(1, 1, 0)$  is :

- Options
1.  $x - 3y - 2z = -2$
  2.  $x + 3y + z = 4$
  3.  $x - y - z = 0$  ✓
  4.  $2x - z = 2$

Question Type : MCQ

Question ID : 41652912779

Option 1 ID : 41652949896

Option 2 ID : 41652949895

Option 3 ID : 41652949894

Option 4 ID : 41652949897

Status : Answered

Chosen Option : 3

Q.8 If  $S_1$  and  $S_2$  are respectively the sets of local minimum and local maximum points of the function,  $f(x) = 9x^4 + 12x^3 - 36x^2 + 25$ ,  $x \in \mathbb{R}$ , then :

- Options
1.  $S_1 = \{-2\}; S_2 = \{0, 1\}$
  2.  $S_1 = \{-1\}; S_2 = \{0, 2\}$
  3.  $S_1 = \{-2, 0\}; S_2 = \{1\}$
  4.  $S_1 = \{-2, 1\}; S_2 = \{0\}$  ✓

Question Type : MCQ

Question ID : 41652912768

Option 1 ID : 41652949853

Option 2 ID : 41652949852

Option 3 ID : 41652949850

Option 4 ID : 41652949851

Status : Answered

Chosen Option : 4

Q.9 If

$$2y = \left( \cot^{-1} \left( \frac{\sqrt{3} \cos x + \sin x}{\cos x - \sqrt{3} \sin x} \right) \right)^2, x \in \left( 0, \frac{\pi}{2} \right)$$

then  $\frac{dy}{dx}$  is equal to :

- Options
1.  $\frac{\pi}{6} - x$
  2.  $2x - \frac{\pi}{3}$
  3.  $x - \frac{\pi}{6}$  ✓

4.  $\frac{\pi}{3} - x$

Question Type : MCQ

Question ID : 41652912766

Option 1 ID : 41652949842

Option 2 ID : 41652949845

Option 3 ID : 41652949844

Option 4 ID : 41652949843

Status : Answered

Chosen Option : 3

Q.10 The magnitude of the projection of the vector  $2\hat{i} + 3\hat{j} + \hat{k}$  on the vector perpendicular to the plane containing the vectors  $\hat{i} + \hat{j} + \hat{k}$  and  $\hat{i} + 2\hat{j} + 3\hat{k}$ , is :

Options

1.  $3\sqrt{6}$

2.  $\sqrt{\frac{3}{2}}$  ✓

3.  $\sqrt{6}$

4.  $\frac{\sqrt{3}}{2}$

Question Type : MCQ

Question ID : 41652912780

Option 1 ID : 41652949898

Option 2 ID : 41652949901

Option 3 ID : 41652949900

Option 4 ID : 41652949899

Status : Answered

Chosen Option : 2

Q.11 Let  $A = \begin{pmatrix} \cos\alpha & -\sin\alpha \\ \sin\alpha & \cos\alpha \end{pmatrix}$ , ( $\alpha \in \mathbb{R}$ ) such that

$$A^{32} = \begin{pmatrix} 0 & -1 \\ 1 & 0 \end{pmatrix}. \text{ Then a value of } \alpha \text{ is :}$$

Options

1. 0

2.  $\frac{\pi}{16}$

3.  $\frac{\pi}{64}$  ✓

4.  $\frac{\pi}{32}$

Question Type : MCQ

Question ID : 41652912759

Option 1 ID : 41652949814

Option 2 ID : 41652949817

Option 3 ID : 41652949815

Option 4 ID : 41652949816

Status : Answered

Chosen Option : 3

Q.12 A point on the straight line,  $3x + 5y = 15$  which is equidistant from the coordinate axes will lie only in :

- Options
1. 1<sup>st</sup> and 2<sup>nd</sup> quadrants ✓
  2. 1<sup>st</sup>, 2<sup>nd</sup> and 4<sup>th</sup> quadrants
  3. 1<sup>st</sup> quadrant
  4. 4<sup>th</sup> quadrant

Question Type : MCQ

Question ID : 41652912774

Option 1 ID : 41652949876

Option 2 ID : 41652949874

Option 3 ID : 41652949875

Option 4 ID : 41652949877

Status : Not Answered

Chosen Option : --

Q.13

$$\int \frac{\sin \frac{5x}{2}}{\sin \frac{x}{2}} dx \text{ is equal to :}$$

(where c is a constant of integration.)

- Options
1.  $x + 2 \sin x + \sin 2x + c$  ✓
  2.  $2x + \sin x + \sin 2x + c$
  3.  $x + 2 \sin x + 2 \sin 2x + c$
  4.  $2x + \sin x + 2 \sin 2x + c$

Question Type : MCQ

Question ID : 41652912769

Option 1 ID : 41652949854

Option 2 ID : 41652949856

Option 3 ID : 41652949855

Option 4 ID : 41652949857

Status : Answered

Chosen Option : 1

Q.14 Let O(0, 0) and A(0, 1) be two fixed points. Then the locus of a point P such that the perimeter of  $\Delta AOP$  is 4, is :

- Options
1.  $8x^2 + 9y^2 - 9y = 18$

2.  $9x^2 - 8y^2 + 8y = 16$   
 3.  $8x^2 - 9y^2 + 9y = 18$   
 4.  $9x^2 + 8y^2 - 8y = 16$  ✓

Question Type : **MCQ**Question ID : **41652912773**Option 1 ID : **41652949873**Option 2 ID : **41652949870**Option 3 ID : **41652949872**Option 4 ID : **41652949871**Status : **Answered**Chosen Option : **4**

Q.15

If  $\cos(\alpha + \beta) = \frac{3}{5}$ ,  $\sin(\alpha - \beta) = \frac{5}{13}$  and

$0 < \alpha, \beta < \frac{\pi}{4}$ , then  $\tan(2\alpha)$  is equal to :

Options

1.  $\frac{21}{16}$   
 2.  $\frac{63}{52}$   
 3.  $\frac{33}{52}$   
 4.  $\frac{63}{16}$  ✓

Question Type : **MCQ**Question ID : **41652912783**Option 1 ID : **41652949910**Option 2 ID : **41652949912**Option 3 ID : **41652949913**Option 4 ID : **41652949911**Status : **Not Answered**Chosen Option : **--**

Q.16

The sum of the squares of the lengths of the chords intercepted on the circle,  $x^2 + y^2 = 16$ , by the lines,  $x + y = n$ ,  $n \in \mathbb{N}$ , where  $\mathbb{N}$  is the set of all natural numbers, is :

- Options 1. **210** ✓  
 2. 105  
 3. 320  
 4. 160

Question Type : **MCQ**Question ID : **41652912775**Option 1 ID : **41652949879**Option 2 ID : **41652949878**

Option 3 ID : 41652949880  
 Option 4 ID : 41652949881  
 Status : Not Answered  
 Chosen Option : --

Q.17  $\lim_{x \rightarrow 0} \frac{\sin^2 x}{\sqrt{2} - \sqrt{1 + \cos x}}$  equals :

- Options
1.  $4\sqrt{2}$  ✓
  2.  $2\sqrt{2}$
  3.  $\sqrt{2}$
  4. 4

Question Type : MCQ  
 Question ID : 41652912765  
 Option 1 ID : 41652949838  
 Option 2 ID : 41652949841  
 Option 3 ID : 41652949840  
 Option 4 ID : 41652949839  
 Status : Answered  
 Chosen Option : 1

Q.18 The mean and variance of seven observations are 8 and 16, respectively. If 5 of the observations are 2, 4, 10, 12, 14, then the product of the remaining two observations is :

- Options
1. 48 ✓
  2. 45
  3. 49
  4. 40

Question Type : MCQ  
 Question ID : 41652912781  
 Option 1 ID : 41652949905  
 Option 2 ID : 41652949904  
 Option 3 ID : 41652949903  
 Option 4 ID : 41652949902  
 Status : Not Answered  
 Chosen Option : --

Q.19 Let A and B be two non-null events such that  $A \subset B$ . Then, which of the following statements is always correct ?

- Options
1.  $P(A|B) \geq P(A)$  ✓
  2.  $P(A|B) = P(B) - P(A)$
  3.  $P(A|B) \leq P(A)$
  4.  $P(A|B) = 1$

Question Type : **MCQ**Question ID : **41652912782**Option 1 ID : **41652949908**Option 2 ID : **41652949909**Option 3 ID : **41652949907**Option 4 ID : **41652949906**Status : **Answered**Chosen Option : **3**

**Q.20** The sum of all natural numbers 'n' such that  $100 < n < 200$  and  $H.C.F. (91, n) > 1$  is :

Options 1. 3203

2. 3221

3. 3121 ✓

4. 3303

Question Type : **MCQ**Question ID : **41652912763**Option 1 ID : **41652949830**Option 2 ID : **41652949831**Option 3 ID : **41652949832**Option 4 ID : **41652949833**Status : **Answered**Chosen Option : **4**

**Q.21** The sum of the co-efficients of all even degree terms in  $x$  in the expansion of  $(x + \sqrt{x^3 - 1})^6 + (x - \sqrt{x^3 - 1})^6$ ,  $(x > 1)$  is equal to :

Options 1. 26

2. 32

3. 24 ✓

4. 29

Question Type : **MCQ**Question ID : **41652912762**Option 1 ID : **41652949827**Option 2 ID : **41652949829**Option 3 ID : **41652949826**Option 4 ID : **41652949828**Status : **Answered**Chosen Option : **1**

**Q.22** The area (in sq.units) of the region  $A = \{(x, y) \in \mathbb{R} \times \mathbb{R} \mid 0 \leq x \leq 3, 0 \leq y \leq 4, y \leq x^2 + 3x\}$  is :

Options 1.  $\frac{26}{3}$ 

2. 8

3.  $\frac{53}{6}$

4.  $\frac{59}{6}$  ✓

Question Type : **MCQ**Question ID : **41652912771**Option 1 ID : **41652949863**Option 2 ID : **41652949862**Option 3 ID : **41652949865**Option 4 ID : **41652949864**Status : **Not Answered**

Chosen Option : --

**Q.23** The greatest value of  $c \in \mathbb{R}$  for which the system of linear equations

$$x - cy - cz = 0$$

$$cx - y + cz = 0$$

$$cx + cy - z = 0$$

has a non-trivial solution, is :

Options 1.  $-1$ 2.  $2$ 3.  $\frac{1}{2}$  ✓4.  $0$ Question Type : **MCQ**Question ID : **41652912760**Option 1 ID : **41652949818**Option 2 ID : **41652949821**Option 3 ID : **41652949820**Option 4 ID : **41652949819**Status : **Answered**Chosen Option : **3**

**Q.24** The sum of the series  $2 \cdot {}^{20}C_0 + 5 \cdot {}^{20}C_1 + 8 \cdot {}^{20}C_2 + 11 \cdot {}^{20}C_3 + \dots + 62 \cdot {}^{20}C_{20}$  is equal to :

Options 1.  $2^{26}$ 2.  $2^{25}$  ✓3.  $2^{24}$ 4.  $2^{23}$ Question Type : **MCQ**Question ID : **41652912764**Option 1 ID : **41652949837**Option 2 ID : **41652949835**Option 3 ID : **41652949834**Option 4 ID : **41652949836**Status : **Answered**



Q.25

If  $f(x) = \frac{2 - x \cos x}{2 + x \cos x}$  and  $g(x) = \log_e x$ ,  
 ( $x > 0$ ) then the value of the integral

$$\int_{-\pi/4}^{\pi/4} g(f(x)) dx \text{ is :}$$

- Options
1.  $\log_e e$
  2.  $\log_e 2$
  3.  $\log_e 1$  ✓
  4.  $\log_e 3$

Question Type : MCQ

Question ID : 41652912770

Option 1 ID : 41652949861

Option 2 ID : 41652949858

Option 3 ID : 41652949860

Option 4 ID : 41652949859

Status : Answered

Chosen Option : 3

Q.26

The contrapositive of the statement "If you are born in India, then you are a citizen of India", is :

- Options
1. If you are not born in India, then you are not a citizen of India.
  2. If you are a citizen of India, then you are born in India.
  3. If you are born in India, then you are not a citizen of India.
  4. If you are not a citizen of India, then you are not born in India. ✓

Question Type : MCQ

Question ID : 41652912785

Option 1 ID : 41652949920

Option 2 ID : 41652949919

Option 3 ID : 41652949918

Option 4 ID : 41652949921

Status : Answered

Chosen Option : 4

Q.27

The sum of the solutions of the equation  
 $|\sqrt{x} - 2| + \sqrt{x}(\sqrt{x} - 4) + 2 = 0$ , ( $x > 0$ )  
 is equal to :

Options

1. 10 ✓
2. 9
3. 12
4. 4

Question Type : **MCQ**  
 Question ID : **41652912758**  
 Option 1 ID : **41652949812**  
 Option 2 ID : **41652949811**  
 Option 3 ID : **41652949813**  
 Option 4 ID : **41652949810**  
 Status : **Answered**  
 Chosen Option : **1**

**Q.28** If  $\alpha$  and  $\beta$  be the roots of the equation  $x^2 - 2x + 2 = 0$ , then the least value of  $n$  for

which  $\left(\frac{\alpha}{\beta}\right)^n = 1$  is :

- Options
1. 5
  2. 4 ✓
  3. 2
  4. 3

Question Type : **MCQ**  
 Question ID : **41652912757**  
 Option 1 ID : **41652949806**  
 Option 2 ID : **41652949807**  
 Option 3 ID : **41652949809**  
 Option 4 ID : **41652949808**  
 Status : **Answered**  
 Chosen Option : **3**

**Q.29** If the tangents on the ellipse  $4x^2 + y^2 = 8$  at the points  $(1, 2)$  and  $(a, b)$  are perpendicular to each other, then  $a^2$  is equal to :

- Options
1.  $\frac{2}{17}$  ✓
  2.  $\frac{4}{17}$
  3.  $\frac{64}{17}$
  4.  $\frac{128}{17}$

Question Type : **MCQ**  
 Question ID : **41652912777**  
 Option 1 ID : **41652949886**  
 Option 2 ID : **41652949888**  
 Option 3 ID : **41652949889**

Option 4 ID : 41652949887

Status : Answered

Chosen Option : 1

**Q.30** Let  $f: [0, 2] \rightarrow \mathbb{R}$  be a twice differentiable function such that  $f''(x) > 0$ , for all  $x \in (0, 2)$ .  
If  $\phi(x) = f(x) + f(2-x)$ , then  $\phi$  is :

- Options
1. decreasing on  $(0, 2)$
  2. increasing on  $(0, 2)$
  3. increasing on  $(0, 1)$  and decreasing on  $(1, 2)$ .
  4. decreasing on  $(0, 1)$  and increasing on  $(1, 2)$ . ✓

Question Type : MCQ

Question ID : 41652912767

Option 1 ID : 41652949849

Option 2 ID : 41652949848

Option 3 ID : 41652949846

Option 4 ID : 41652949847

Status : Answered

Chosen Option : 3